Tarapacáite K_2CrO_4

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Crystal Data: Orthorhombic. Point Group: $2/m \ 2/m \ 2/m$. Rarely as thick tabular $\{001\}$ crystals; typically disseminated granular. Twinning: As pseudohexagonal trillings on $\{011\}$.

Physical Properties: Cleavage: On $\{010\}$ and $\{100\}$, distinct. Hardness = n.d. D(meas.) = 2.74 (synthetic). D(calc.) = 2.735 Soluble in H_2O .

Optical Properties: Transparent. Color: Bright canary-yellow. Optical Class: Biaxial (-). Orientation: X=c; Y=a; Z=b. Dispersion: r>v, weak. $\alpha=1.687$ $\beta=1.722$ $\gamma=1.731$ $2V(\text{meas.})=52^{\circ}$

Cell Data: Space Group: Pnam (synthetic). a=7.662(1) b=10.391(1) c=5.919(1) Z=4

X-ray Powder Pattern: Synthetic. 3.078 (100), 2.988 (75), 2.960 (40), 2.479 (25), 2.286 (25), 2.599 (22), 2.570 (21)

Chemistry: (1) Analyses of natural material are lacking; identification depends on coincidence of optical and X-ray data with synthetic material.

Occurrence: A minor accessory mineral in nitrate deposits.

Association: Lopezite, dietzeite.

Distribution: In Chile, in nitrate caliche deposits throughout the Atacama Desert, as at the Oficina Maria Elena, near Tocopilla, Antofagasta, and Zapiga, Tarapacá.

Name: For its first noted occurrence in the Tocopilla Pampa, 170 km north of Antofagasta, Tarapacá, Chile.

References: (1) Palache, C., H. Berman, and C. Frondel (1951) Dana's system of mineralogy, (7th edition), v. II, 644-645. (2) Pistorius, C.W.F.T. (1962) Phase relations of some potassium compounds to very high pressures. Z. Physik. Chem., 35, 109–121. (3) McGinnety, J.A. (1972) Redetermination of the structures of potassium sulphate and potassium chromate: the effect of electrostatic crystal forces upon observed bond lengths. Acta Cryst., 28, 2845–2858. (4) Toriumi, K. and Y. Saito (1978) Electron-density distribution in crystals of α –K₂CrO₄. Acta Cryst., 34, 3149–3156.