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Crystal Data: Monoclinic. Point Group: 2/m (synthetic). As euhedral to anhedral crystals, to 1 mm, in sugary-textured to fine-grained aggregates.

**Physical Properties:** Hardness = n.d. D(meas.) = 1.713 (synthetic). D(calc.) = 1.711 Soluble in  $H_2O$ .

Optical Properties: Semitransparent. Color: [White.] Optical Class: Biaxial (+).  $\alpha = 1.432-1.435$   $\beta = \text{n.d.}$   $\gamma = 1.450-1.488$  2V(meas.) = Small.

Cell Data: Space Group: C2/c (synthetic). a = 11.189(8) b = 16.474(14) c = 13.576(9)  $\beta = 112^{\circ}50(2)'$  Z = 8

**X-ray Powder Pattern:** Furnace Creek district, California, USA. 4.60 (10), 3.30 (8), 3.20 (7), 3.74 (5), 3.18 (5), 4.29 (3), 3.56 (3)

**Chemistry:** (1) Identification depends on identity of X-ray powder pattern and optical data with synthetic material.

Occurrence: Formed in pipes at a reduction works for borates from hot spring lagoons, between 32° C and 60° C (Larderello, Italy); deposited in an arid climate in surficial debris above weathering borate-bearing veins and in beds of saline tuffaceous siltstone (Furnace Creek district, California, USA).

**Association:** Borax, thénardite (Larderello, Italy); halite, thénardite (Furnace Creek district, California, USA).

**Distribution:** From Larderello, Val di Cecina, Tuscany, Italy. At several localities in the Furnace Creek district, Death Valley, Inyo Co., California, USA.

Name: Honors Professor Umberto Sborgi (1883–1955), Italian chemist, University of Milan, Milan, Italy, a worker in the system Na<sub>2</sub>O-B<sub>2</sub>O<sub>3</sub>-H<sub>2</sub>O.

**Type Material:** University of Florence, Florence, Italy, 16801/702.

References: (1) Cipriani, C. (1957) Un nuovo minerale fra i prodotti boriferi di Larderello. Atti Rend. Accad. Lincei, 22, 519–525 (in Italian). (2) (1958) Amer. Mineral., 43, 378 (abs. ref. 1). (3) McAllister, J.F. (1961) Sborgite in the Furnace Creek area, California. U.S. Geol. Surv. Prof. Paper 424, B299–B301. (4) Merlino, S. and F. Sartori (1972) The crystal structure of sborgite, NaB<sub>5</sub>O<sub>6</sub>(OH)<sub>4</sub>.3H<sub>2</sub>O. Acta Cryst., 28, 3559–3567.