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Crystal Data: Hexagonal. Point Group: 3m. Steep trigonal rhombohedra $\{03\overline{3}2\}$, modified by $\{10\overline{1}0\}$, $\{10\overline{1}1\}$, $\{01\overline{1}2\}$, $\{0001\}$, to 2 mm; may form polycrystals with galeite.

Physical Properties: Fracture: Conchoidal. Tenacity: Brittle. Hardness = $3.5 \, \mathrm{D(meas.)} = 2.612 - 2.616 \, \mathrm{D(calc.)} = 2.619 \, \mathrm{Slowly}$ soluble in $\mathrm{H}_2\mathrm{O}$.

Optical Properties: Transparent to translucent with clay inclusions. *Color:* Colorless; colorless in transmitted light. *Luster:* Vitreous.

Optical Class: Uniaxial (+). $\omega = 1.440$ $\epsilon = 1.445$

Cell Data: Space Group: P31m. a = 12.197(4) c = 19.259(11) Z = 3

X-ray Powder Pattern: Searles Lake, California, USA. 2.76 (10), 3.52 (8), 3.79 (7), 1.760 (7), 3.01 (6), 2.58 (6), 4.44 (2)

Chemistry:

	(1)	(2)	(3)
SO_3	42.79	42.96	42.95
Na_2O	49.51	49.54	49.88
K_2O	0.16		
\mathbf{F}^{-}	8.30	8.55	8.74
Cl	3.53	2.76	2.72
$-\mathcal{O} = (\mathcal{F}, \mathcal{Cl})_2$	4.29	4.22	4.29
Total	[100.00]	99.59	100.00

(1) Searles Lake, California, USA; recalculated to oxides from an original total of 98.95%, after deduction of (Fe, Al)₂O₃ 0.15%, CaO 0.30%, loss on ignition 0.90%, insoluble 0.20%. (2) Do.; by electron microprobe, average of two sets of analyses, recalculated to oxides from an original total of 99.61%. (3) $Na_{21}(SO_4)_7F_6Cl$.

Occurrence: In a playa lake deposit (Searles Lake, California, USA); associated with an intrusive alkalic gabbro-syenite complex (Mont Saint-Hilaire, Canada).

Association: Galeite, gaylussite, tychite, pirssonite, thénardite, northupite, trona, hanksite, calcite (Searles Lake, California, USA); pectolite, shortite, kogarkoite, phlogopite, arfvedsonite, leucophanite, leucophanite (Mont Saint-Hilaire, Canada).

Distribution: In the USA, from Searles Lake, San Bernardino Co., California. At Mont Saint-Hilare, Quebec, Canada. From the Nagyvisnyó evaporite deposit, Bükk Mountains, Hungary.

Name: To honor Dr. John Frank Schairer (1904–1970), American physical chemist, Carnegie Geophysical Laboratory, Washington, D.C., USA, who studied the system $Na_2SO_4-NaF-NaCl-H_2O$.

Type Material: National Museum of Natural History, Washington, D.C., USA, 96437, 96439, 112736.

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