**Crystal Data:** Orthorhombic (synthetic). Point Group: 2/m 2/m 2/m. Small indistinct flat crystals.

**Physical Properties:** Hardness = n.d. D(meas.) = n.d. D(calc.) = 1.83 Soluble in  $H_2O$ , forming a struvite residue.

**Optical Properties:** Transparent. Color: Colorless. Optical Class: Biaxial (+) (synthetic). Orientation: Z = a; OAP = {001}.  $\alpha = 1.508(2)$  $\beta = 1.515(2)$   $\gamma = 1.523(2)$  2V(meas.) = ~90° 2V(calc.) = 86°

Cell Data: Space Group: Pbca (probable; synthetic). a = 11.47 b = 23.63 c = 8.62 Z = 8

## X-ray Powder Pattern: Synthetic.

5.94 (100), 2.97 (43), 5.21 (37), 2.80 (29), 3.02 (22), 3.46 (21), 4.31 (19)

Chemistry:		(1)	(2)	
	$P_2O_5$	43.88	43.76	
	FeO	0.20		
	MnO	0.05		
	MgO	12.17	12.42	
	$(NH_4)_2O$	16.15	16.05	
	H <sub>2</sub> O	27.55	27.77	
	Total	100.00	100.00	
(1) Skipton lave tube eares	$\Lambda_{\rm ustralia}$ (2)	(NH) Ma	PO OH)	1H

(1) Skipton lava tube caves, Australia. (2)  $(NH_4)_2Mg(PO_3OH)_2 \cdot 4H_2O$ .

**Occurrence:** Rarely formed in drier portions of bat guano, by the reaction of  $NH_4$  from guano with Mg in surrounding rocks.

Association: Struvite, newberyite (Skipton lava tube caves, Australia).

**Distribution:** In the Skipton lava tube caves, 40 km southwest of Ballarat, Victoria, Australia. From Chaos Cave, Transvaal, South Africa.

Name: To honor Professor Arnulf Schertel (1841–1902), mineralogist, Freiberg Mining Academy, Freiberg, Germany.

**References:** (1) Palache, C., H. Berman, and C. Frondel (1951) Dana's system of mineralogy, (7th edition), v. II, 699. (2) Frazier, A.W., J.R. Lehr, and J.P. Smith (1963) The magnesium phosphates hannayite, schertelite and bobierrite. Amer. Mineral., 48, 635–641. (3) Catti, M. and M. Franchini-Angela (1976) Hydrogen bonding in the crystalline state. Structure of  $Mg_3(NH_4)_2(HPO_4)_4 \cdot 4H_2O$  (hannayite), and crystal-chemical relationships with schertelite and struvite. Acta Cryst., 32, 2842–2848.