Crystal Data: Monoclinic. *Point Group*: 2/*m*. As needle-like crystals, to 0.5 mm, elongated

parallel to [001] with dominant $\{100\}$ and $\{110\}$; additional forms $\{111\}$, $\{221\}$, $\{131\}$, $\{\overline{1}01\}$.

Physical Properties: *Cleavage*: None. *Fracture*: Uneven. *Tenacity*: Brittle. Hardness = 3.5 D(meas.) = 3.15(3) D(calc.) = 3.17

Optical Properties: Transparent. *Color*: Colorless. *Streak*: White. *Luster*: Vitreous. *Optical Class*: Biaxial (-). $\alpha = 1.5884(10)$ $\beta = 1.6445$ (calc.) $\gamma = 1.6455(10)$ $2V = 15.0(5)^{\circ}$ *Orientation*: $Y \land c = 26^{\circ}$ (in acute β); Z = b.

Cell Data: Space Group: C2/c. a = 19.045(3) b = 9.320(2) c = 6.525(1) $\beta = 92.73(2)^{\circ}$ Z = 4

X-ray Powder Pattern: Skorpion mine, Lüderitz district, Karas region, south-western Namibia. Intensities corrected to remove effects of preferred orientation. 3.170 (100), 2.788 (67), 3.014 (54), 9.501 (53), 3.063 (42), 5.238 (30), 2.582 (21)

Chemistry:		(1)	(2)
	CaO	30.89	30.42
	ZnO	28.83	29.43
	P_2O_5	25.49	25.67
	CO_2 (calc)	7.96	7.96
	H_2O (calc)	6.52	6.52
	Total	99.69	100.00

1) Skorpion mine, Lüderitz district, Karas region, south-western Namibia; average of 17 electron microprobe analyses, anionic groups confirmed by IR, H_2O and CO_2 calculated, corresponding to $Ca_{3.05} Zn_{1.96}(PO_4)_{1.99}(CO_3)_{1.00}(OH)_{2.06} \cdot 0.98H_2O$. (2) $Ca_3Zn_2(PO_4)_2CO_3(OH)_2 \cdot H_2O$.

Occurrence: A secondary mineral in an oxidized non-sulfide zinc deposit formed by weathering of sediment- and volcanic-hosted disseminated sulfide minerals.

Association: Tarbuttite, hydrozincite, gypsum.

Distribution: Skorpion zinc mine, Lüderitz district, Karas region, south-western Namibia.

Name: For the locality that produced the first specimens.

Type Material: Mineralogical Institute, University of Bochum, Germany, IMA 2005-010.

References: (1) Krause, W., H. Effenberger, H.-J. Bernhardt, and O. Medenbach (2008) Skorpionite, $Ca_3Zn_2(PO_4)_2CO_3(OH)_2 \cdot H_2O$, a new mineral from Namibia: description and crystal structure. Eur. J. Mineral., 20, 271–280. (2) (2009) Amer. Mineral., 94, 403 (abs. ref. 1).